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Key indicators

Single-crystal X-ray study

T = 300 K

Mean $\sigma(\text{C}-\text{C}) = 0.005 \text{ \AA}$

R factor = 0.037

wR factor = 0.090

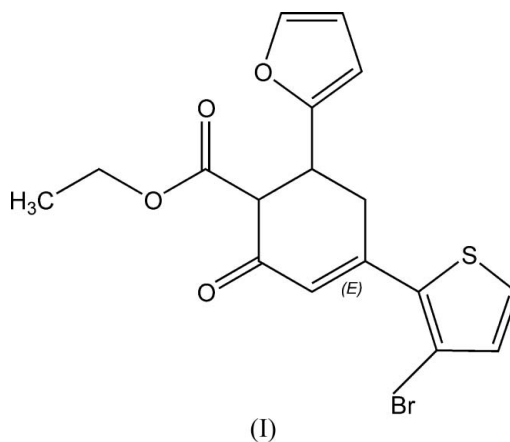
Data-to-parameter ratio = 16.3

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.(8*RS*,9*SR*)-Ethyl 4-(3-bromothiien-2-yl)-6-(2-furyl)-2-oxocyclohex-3-ene-1-carboxylateThe title compound, $\text{C}_{17}\text{H}_{15}\text{BrO}_4\text{S}$, was synthesized from (2*E*)-1-(3-bromo-2-thienyl)-3-(2-furyl)prop-2-en-1-one and ethyl acetoacetate in an ethanol solution. Single crystals were obtained from an ethyl acetate/hexane mixture. The crystal packing is stabilized by van der Waals forces.

Received 15 November 2006

Accepted 5 December 2006

Comment

Chalcones and the corresponding heterocyclic analogs are valuable intermediates in organic synthesis (Dhar, 1981) and exhibit a multitude of biological activities (Dimmock *et al.*, 1999). From a chemical point of view, an important feature of chalcones and their heterocyclic analogs is their ability to act as activated unsaturated systems in conjugated addition reactions of carbanions in the presence of basic catalysts (House, 1972). This type of reaction may be exploited with the goal of obtaining highly functionalized cyclohexene derivatives (Tabba *et al.*, 1995), but is more commonly used for the preparation of 3,5-diaryl-6-carbomethoxycyclohexenones *via* the Michael addition of ethyl acetoacetate. Cyclohexenones are efficient synthons in building spiro compounds (Padmavathi, Sharmila, Somashekara Reddy *et al.*, 2001) or intermediates in the synthesis of benzisoxazoles or carbazole derivatives (Padmavathi *et al.*, 1999, 2000; Padmavathi, Sharmila, Balaiah *et al.*, 2001). In view of the importance of these derivatives, a new compound, ethyl 3-(3-bromothiien-2-yl)-6-(2-furyl)-2-oxocyclohex-3-ene-1-carboxylate, $\text{C}_{17}\text{H}_{15}\text{BrO}_4\text{S}$, (I), has been prepared and its crystal structure is reported here.

Compound (I) was prepared by the cyclocondensation of ethyl acetoacetate with a chalcone, leading to the generation of two chiral centers at C8 and C9 (Fig. 1). As the reaction is not stereoselective, both configurations of the chiral carbon atoms are expected to be found in (I), which would result in a

mixture of four diastereomers. In the title structure, a pair of 8*S*,9*R* and 8*R*,9*S* enantiomers is found. No attempt has been made to separate the diastereomers, and the crystal studied was grown from the mixture on recrystallization.

The bond lengths and angles of (I) are unexceptional. The cyclohexene ring adopts an envelope conformation, with C9 in the flap position. The dihedral angle between the thiophene and furan rings is 58.95 (14)°. One methyl group, C17, appears to exhibit slight disorder; however, refinement using a split model did not yield a satisfactory structural model. The crystal packing is stabilized by van der Waals forces.

Experimental

(2*E*)-1-(3-Bromo-2-thienyl)-3-(2-furyl)prop-2-en-1-one (1.42 g, 5 mmol) and ethyl acetoacetate (0.65 g, 5 mmol) were refluxed for 3 h in 15 ml of ethanol in the presence of 0.8 ml of 10% NaOH. The reaction mixture was cooled to 300 K. The product formed was filtered off and recrystallized from methanol. Crystals were grown from a (1:1) ethyl acetate–hexane solvent mixture (yield: 57%; m.p. 366–368 K). Analysis found (calculated) for C₁₇H₁₅BrO₄S (%): C: 51.66 (51.64), H: 3.83 (3.81), S: 8.11 (8.09).

Crystal data

C ₁₇ H ₁₅ BrO ₄ S	Z = 4
<i>M_r</i> = 395.26	<i>D_x</i> = 1.582 Mg m ⁻³
Monoclinic, <i>P</i> 2 ₁ / <i>c</i>	Mo <i>K</i> α radiation
<i>a</i> = 16.5539 (14) Å	<i>μ</i> = 2.62 mm ⁻¹
<i>b</i> = 11.1125 (8) Å	<i>T</i> = 300 K
<i>c</i> = 9.1650 (5) Å	Irregular block, colourless
<i>β</i> = 100.098 (6)°	0.37 × 0.35 × 0.22 mm
<i>V</i> = 1659.8 (2) Å ³	

Data collection

Bruker–Nonius KappaCCD diffractometer	17471 measured reflections
<i>φ</i> scans	3390 independent reflections
Absorption correction: numerical (<i>HABITUS</i> ; Herrendorf & Bärnighausen, 1997)	2655 reflections with <i>I</i> > 2σ(<i>I</i>)
<i>T_{min}</i> = 0.600, <i>T_{max}</i> = 0.805	<i>R_{int}</i> = 0.060
	<i>θ_{max}</i> = 26.5°

Refinement

Refinement on <i>F</i> ²	$w = 1/[\sigma^2(F_o^2) + (0.025P)^2 + 1.728P]$
$R[F^2 > 2\sigma(F^2)] = 0.037$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.090$	(Δ/σ) _{max} < 0.001
<i>S</i> = 1.17	$\Delta\rho_{max} = 0.31 \text{ e } \text{Å}^{-3}$
3390 reflections	$\Delta\rho_{min} = -0.45 \text{ e } \text{Å}^{-3}$
208 parameters	
H-atom parameters constrained	

H atoms were placed at calculated positions and refined using a riding model; *U*_{iso}(H) = 1.2*U*_{eq}(carrier atom), *U*_{iso}(H) = 1.5*U*_{eq}(methyl) and C–H = 0.93–0.98 Å.

Data collection: *COLLECT* (Nonius, 1999); cell refinement: *DIRAX/LSQ* (Duisenberg, 1992); data reduction: *EVALCCD* (Duisenberg *et al.*, 2003); program(s) used to solve structure:

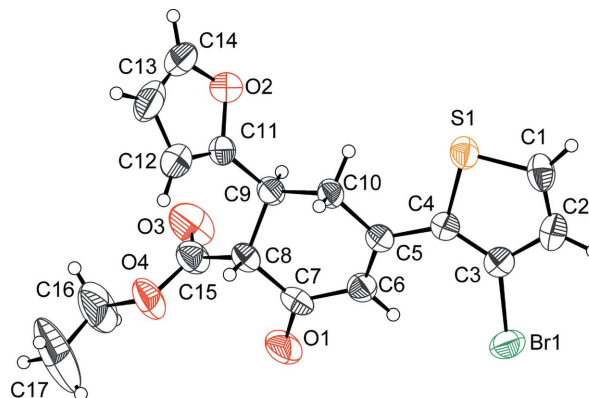


Figure 1

The molecular structure of (I). Displacement ellipsoids are drawn at the 50% probability level.

SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *maXus* (Mackay *et al.*, 1999).

One of the authors (BVA) is grateful to Mangalore University for research facilities. The Swedish Research Council (VR) is acknowledged for providing funding for the single-crystal diffractometer.

References

- Brandenburg, K. (2006). *DIAMOND*. Release 3.1d. Crystal Impact GbR, Bonn, Germany.
- Dhar, D. N. (1981). *The Chemistry of Chalcones and Related Compounds*. New York: Wiley-Interscience.
- Dimmock, J. R., Elias, D. W., Beazely, M. A. & Kandepu, N. M. (1999). *Curr. Med. Chem.*, **6**, 1125–1150.
- Duisenberg, A. J. M. (1992). *J. Appl. Cryst.* **25**, 92–96.
- Duisenberg, A. J. M., Kroon-Batenburg, L. M. J. & Schreurs, A. M. M. (2003). *J. Appl. Cryst.* **36**, 220–229.
- Herrendorf, W. & Bärnighausen, H. (1997). *HABITUS*. University of Karlsruhe, Germany.
- House, H. O. (1972). *Modern Synthetic Reactions*, 2nd ed., p. 595. Menlo Park, California: W. A. Benjamin.
- Mackay, S., Gilmore, C. J., Edwards, C., Stewart, N. & Shankland, K. (1999). *maXus*. Bruker–Nonius, The Netherlands, MacScience, Japan, and The University of Glasgow, Scotland.
- Nonius (1999). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Padmavathi, V., Mohana Reddy, B. J., Balaiah, A., Venugopal Reddy, K. & Bhaskar Reddy, D. (2000). *Molecules*, **5**, 1281–1286.
- Padmavathi, V., Sharmila, K., Balaiah, A., Somashekara Reddy, A. & Bhaskar Reddy, D. (2001). *Synth. Commun.* **31**, 2119–2126.
- Padmavathi, V., Sharmila, K., Padmaja, A. & Bhaskar Reddy, D. (1999). *Heterocycl. Commun.* **5**, 451–456.
- Padmavathi, V., Sharmila, K., Somashekara Reddy, A. & Bhaskar Reddy, D. (2001). *Indian J. Chem. Sect B*, **40**, 11–14.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Tabba, H. D., Yousef, N. M. & Alarab, M. M. (1995). *Collect. Czech. Chem. Commun.* **60**, 594–604.