Acta Crystallographica Section E

Structure Reports Online

ISSN 1600-5368

2-Chloro-4-methylpyridin-3-amine

H. S. Yathirajan, S. Bindya, A. M. Vijesh, B. Narayana and Michael Boltec*

^aDepartment of Studies in Chemistry, University of Mysore, Manasagangotri, Mysore 570 006, India, ^bDepartment of Chemistry, Mangalore University, Mangalagangotri 574 199, India, and ^cInstitut für Anorganische Chemie, J. W. Goethe-Universität Frankfurt, Max-von-Laue-Strasse 7, 60438 Frankfurt/Main, Germany

Correspondence e-mail: bolte@chemie.uni-frankfurt.de

Key indicators

Single-crystal X-ray study T = 173 KMean $\sigma(\text{C-C}) = 0.002 \text{ Å}$ R factor = 0.034 wR factor = 0.093Data-to-parameter ratio = 13.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Geometric parameters of the title compound, $C_6H_7ClN_2$, are in the usual ranges. The molecular structure shows one intramolecular $N-H\cdots Cl$ contact and the crystal packing is stabilized by an intermolecular $N-H\cdots N$ hydrogen bond.

Received 10 January 2007 Accepted 10 January 2007

Comment

Pyridine is an important structural unit found in many known therapeutic agents (Proudfoot *et al.*, 1995). Pyridine and its derivatives are important in industrial organic chemistry as fundamental building blocks (Sherman, 2004). Many pyridinyl thiazoles have proved to possess a wide range of biological activities such as cardiotonic, anti-asthmatic, anti-inflammatory and also shown to be selective inhibitors of cytochrome P-450 2A6 (Denton *et al.*, 2005). Pyridine derivatives are known for their cardiac effects (Schoepke & Shideman, 1962). In view of the importance of pyridine derivatives, the crystal structure of the title compound, (I), is reported.

A perspective view of (I) is shown in Fig. 1. Bond lengths and angles can be regarded as normal (Cambridge Structural Database, Version 5.27, November 2005 update, August 2006;

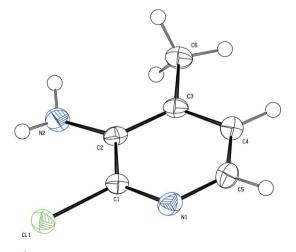


Figure 1The molecular structure of the title compound with the atom numbering; displacement ellipsoids are drawn at the 50% probability level.

© 2007 International Union of Crystallography All rights reserved

Yathirajan et al. • C₆H₇ClN₂

organic papers

MOGUL Version 1.1; Allen, 2002; Bruno et al., 2004). As expected the molecule is planar (r.m.s. deviation for all non-H atoms 0.012 Å). The molecular conformation is characterized by an N-H \cdots Cl contact and the crystal packing is stabilized by an N-H \cdots N hydrogen bond, forming chains along the c axis (Table 1).

Experimental

A pure sample of the title compound was obtained from Srides Arco Laboratory, Mangalore, India. The sample was crystallized from acetone by slow evaporation (m.p. 333–335 K).

Crystal data

C ₆ H ₇ ClN ₂	Z = 4
$M_r = 142.59$	$D_x = 1.440 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
a = 3.9877 (8) Å	$\mu = 0.48 \text{ mm}^{-1}$
b = 12.8468 (15) Å	T = 173 (2) K
c = 12.8408 (19) Å	Rod, colourless
$\beta = 90.872 \ (14)^{\circ}$	$0.48 \times 0.21 \times 0.20 \text{ mm}$
$V = 657.75 (18) \text{ Å}^3$	

Data collection

Stoe IPDS-II two-circle	3879 measured reflections
diffractometer	1226 independent reflections
ω scans	1109 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan	$R_{\rm int} = 0.050$
(MULABS; Spek, 2003; Blessing,	$\theta_{\rm max} = 25.6^{\circ}$
1995)	
$T_{\min} = 0.802, T_{\max} = 0.910$	

Refinement

Refinement on F^2	$w = 1/ \sigma^2(F_0^2) + (0.0536P)^2$
$R[F^2 > 2\sigma(F^2)] = 0.034$	+ 0.1927P]
$wR(F^2) = 0.093$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.08	$(\Delta/\sigma)_{\rm max} < 0.001$
1226 reflections	$\Delta \rho_{\text{max}} = 0.33 \text{ e Å}^{-3}$
92 parameters	$\Delta \rho_{\min} = -0.24 \text{ e Å}^{-3}$
H atoms treated by a mixture of	Extinction correction: SHELXL97
independent and constrained	Extinction coefficient: 0.034 (8)
refinement	

Table 1 Hydrogen-bond geometry (Å, °).

D $ H$ $\cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
$ \begin{array}{c} N2 - H2A \cdots Cl1 \\ N2 - H2B \cdots N1^{i} \end{array} $	0.86 (3) 0.84 (2)	2.55 (3) 2.28 (2)	2.9796 (17) 3.089 (2)	111.9 (19) 162 (2)
Symmetry code: (i) x	z , $-y + \frac{1}{2}$, $z + \frac{1}{2}$.			

H atoms were found in a difference map, but those bonded to C were refined using a riding model, with C-H = 0.95 Å for aromatic or C-H = 0.98 Å for methyl H atoms. $U_{\rm iso}({\rm H})$ values were set at $1.2U_{\rm eq}({\rm C})$ or $1.5U_{\rm eq}({\rm methyl}\ {\rm C})$. The methyl group was allowed to rotate but not to tip. H atoms bonded to nitrogen were refined freely.

Data collection: *X-AREA* (Stoe & Cie, 2001); cell refinement: *X-AREA*; data reduction: *X-AREA*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *PLATON* and *SHELXL97*.

AMV thanks Mangalore University for access to research facilities.

References

Allen, F. H. (2002). Acta Cryst. B58, 380–388.
Blessing, R. H. (1995). Acta Cryst. A51, 33–38.
Bruno, I. J., Cole, J. C., Kessler, M., Luo Jie, Motherwell, W. D. S., Purkis, L. H., Smith, B. R., Taylor, R., Cooper, R. I., Harris, S. E. & Orpen, A. G. (2004). J. Chem. Inf. Comput. Sci. 44, 2133–2144.
Denton, T. T., Zhang, X. & Cashman, J. R. (2005). J. Med. Chem. 48, 224–239.

Proudfoot, J. R., Patel, U. R., Kapadia, S. R. & Hargrave, K. D. (1995). *J. Med. Chem.* 38, 1406–1410.

Schoepke, H. G. & Shideman, F. E. (1962). *J. Pharmacol. Exptl Ther.* **135**, 358–366.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. Univ. of Göttingen, Germany.

Sherman, A. R. (2004). *Encyclopedia of Reagents for Organic Synthesis*, edited by L. Paquette. New York: J. Wiley & Sons.

Spek, A. L. (2003). J. Appl. Cryst. 36, 7-13.

Stoe & Cie (2001). X-AREA. Stoe & Cie, Darmstadt, Germany.