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Key indicators

Single-crystal X-ray study T = 273 KMean $\sigma(\text{C-C}) = 0.004 \text{ Å}$ R factor = 0.033 wR factor = 0.086Data-to-parameter ratio = 17.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1-(3-Bromo-2-thienyl)-3-[4-(dimethylamino)-phenyl]prop-2-en-1-one

The title compound, $C_{15}H_{14}BrNOS$, crystallizes with two molecules in the asymmetric unit. One of the two molecules forms dimers held together by weak $C-H\cdots O$ interactions. The twist angles between the thienyl and benzene rings are 2.70 (16) and 4.76 (18)°, smaller than usually observed in chalcone derivatives.

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Comment

Among several organic compounds reported as having nonlinear optical (NLO) properties, chalcone derivatives are notable materials for their excellent blue-light transmittance and good crystallizability. They provide a necessary configuration to show NLO properties with two planar rings connected through a conjugated double bond (Goto *et al.*, 1991; Uchida *et al.*, 1998; Tam *et al.*, 1989; Indira *et al.*, 2002). In a continuation of our work on chalcones (Butcher *et al.*, 2006, 2006*a,b*), the present paper reports the crystal structure of a newly synthesized chalcone, (I).

Unlike many other chalcone derivatives, which tend to be chiral due to the twist of the two planar rings, the title compound, (I), crystallizes in the centrosymmetric space group $P\overline{1}$, with two molecules in the asymmetric unit. The dihedral angle between the thienyl and phenyl planes in the two molecules are 2.70 (16) and 4.76 (18)°, smaller than the value usually observed in chalcone derivatives (Butcher *et al.*, 2006, 2006*a,b*). One of the two molecules in the asymmetric unit forms dimers which are linked by weak $C-H\cdots O$ interactions (Fig. 2 and Table 1). The metrical parameters of the thienyl and phenyl rings and the chalcone backbone in both molecules are within the normal ranges (Allen, 2002).

Experimental

2-Acetyl-3-bromothiophene (10 g, 0.048 mol) in methanol (50 ml) was mixed with 4-dimethylaminobenzaldehyde (7.16 g, 0.048 mol) and the mixture was treated with 10 ml of 30% potassium hydroxide solution at 278 K. The reaction mixture was then brought to room temperature and stirred for 4 h. The solid which precipitated was filtered and washed with water, dried and recrystallized from acetone (yield 70%; m.p. 371 K).

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Crystal data

C ₁₅ H ₁₄ BrNOS	$V = 1461.51 (9) \text{ Å}^3$
$M_r = 336.24$	Z = 4
Triclinic, $P\overline{1}$	$D_x = 1.528 \text{ Mg m}^{-3}$
a = 9.6866 (3) Å	Mo $K\alpha$ radiation
b = 12.6923 (4) Å	$\mu = 2.95 \text{ mm}^{-1}$
c = 12.8555 (5) Å	T = 273 (2) K
$\alpha = 94.444 \ (1)^{\circ}$	Plate, colourless
$\beta = 101.673 \ (1)^{\circ}$	$0.30 \times 0.25 \times 0.10 \text{ mm}$
$\gamma = 107.322 \ (1)^{\circ}$	

Data collection

Bruker APEX-2 CCD area-detector diffractometer φ and ω scans Absorption correction: multi-scan (SADABS; Sheldrick, 1996) $T_{\min} = 0.271, T_{\max} = 0.590$ (expected range = 0.342–0.745)

13636 measured reflections 6034 independent reflections 4472 reflections with $I > 2\sigma(I)$ $R_{\rm int} = 0.020$ $\theta_{\rm max} = 26.6^{\circ}$

Refinement

 $\begin{array}{lll} \mbox{Refinement on } F^2 & & & & & & & & & \\ R[F^2 > 2\sigma(F^2)] = 0.033 & & & & & & & \\ wR(F^2) = 0.086 & & & & & & & \\ S = 1.02 & & & & & & & \\ 6034 \ \mbox{reflections} & & & & & \\ 347 \ \mbox{parameters} & & & & & \\ \mbox{H-atom parameters constrained} & & & & & & \\ \end{array} \qquad \begin{array}{ll} w = 1/[\sigma^2(F_{\rm o}^2) + (0.0395P)^2 \\ & + 0.3574P] \\ \mbox{where } P = (F_{\rm o}^2 + 2F_{\rm c}^2)/3 \\ \mbox{(Δ/σ)}_{\rm max} = 0.001 \\ \mbox{$\Delta\rho_{\rm max} = 0.54 \ e\ \mathring{A}^{-3}$} \\ \mbox{$\Delta\rho_{\rm min} = -0.43 \ e\ \mathring{A}^{-3}$} \end{array}$

Table 1 Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
$ \begin{array}{c} C7A - H7AA \cdots O1A^{i} \\ C1B - H1BA \cdots O1A^{ii} \end{array} $	0.93	2.52	3.404 (3)	159
	0.93	2.43	3.341 (4)	168

Symmetry codes: (i) -x - 1, -y + 2, -z; (ii) x + 1, y, z.

All H atoms were initially located in a difference Fourier map. The methyl H atoms were then constrained to an ideal geometry, with C—H = 0.98 Å and $U_{\rm iso}({\rm H}) = 1.5 U_{\rm eq}({\rm C})$, but each group was allowed to rotate freely about its C—C bond. All other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C—H = 0.93–0.96 Å and $U_{\rm iso}({\rm H}) = 1.2 U_{\rm eq}({\rm C})$.

Data collection: *APEX2* (Bruker, 2006); cell refinement: *APEX2*; data reduction: *APEX2*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2000); software used to prepare material for publication: *SHELXTL*.

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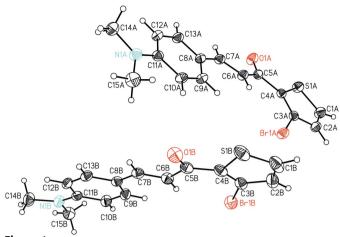


Figure 1The asymmetric unit of of (I) containing two molecules and showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probablity level.

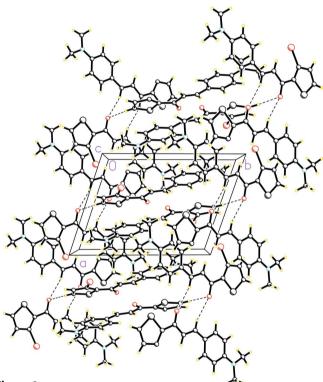


Figure 2 The molecular packing of (I), viewed down the c axis. Dashed lines indicate hydrogen bonds.

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