

4-[(1*E*)-Benzylideneamino]-3-methyl-2,4-dihydro-1*H*-1,2,4-triazole-5-thioneH. S. Yathirajan,<sup>a</sup> B. K. Sarojini,<sup>b</sup>  
B. Narayana,<sup>c</sup> K. Sunil<sup>c</sup> and  
Michael Bolte<sup>d\*</sup><sup>a</sup>Department of Studies in Chemistry, University of Mysore, Manasagangotri, Mysore 570 006, India, <sup>b</sup>Department of Chemistry, P. A. College of Engineering, Nadupadavu, Mangalore 574 153, India, <sup>c</sup>Department of Chemistry, Mangalore University, Mangalagangotri 574 199, India, and <sup>d</sup>Institut für Anorganische Chemie, J. W. Goethe-Universität Frankfurt, Max-von-Laue-Strasse 7, 60438 Frankfurt/Main, GermanyCorrespondence e-mail:  
bolte@chemie.uni-frankfurt.de

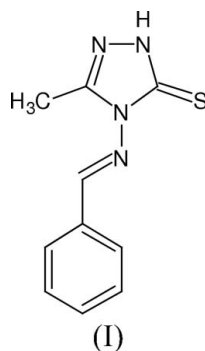
## Key indicators

Single-crystal X-ray study  
*T* = 173 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$   
*R* factor = 0.037  
*wR* factor = 0.101  
Data-to-parameter ratio = 13.8For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The acyclic C=N double bond of the title compound, C<sub>10</sub>H<sub>10</sub>N<sub>4</sub>S, is *trans* configured. The molecule is almost planar. The angle between the two rings is just 10.25 (7)°. The crystal packing is stabilized by N—H···S hydrogen bonds and  $\pi$ – $\pi$  interactions.

## Comment

Condensed 1,2,4-triazoles are biologically important compounds and are used as starting materials for the synthesis of many heterocycles. Apart from its extensive chemical significance, the 1,2,4-triazole nucleus is found to be associated with diverse medicinal properties and is incorporated in a wide variety of therapeutically interesting drugs (Yathirajan *et al.*, 2005). Recently, some new triazole derivatives have been synthesized as possible anticonvulsants, antidepressants, tranquilizers and plant-growth regulators. Substituted 1,2,4-triazoles have been actively studied as bridging ligands, coordinating through their vicinal N atoms. Complexes containing 1,2,4-triazole ligands possess specific magnetic properties. Other 1,2,4-triazole derivatives have anti-inflammatory activities and some are antifungal agents (Ambalavanan *et al.*, 2003).

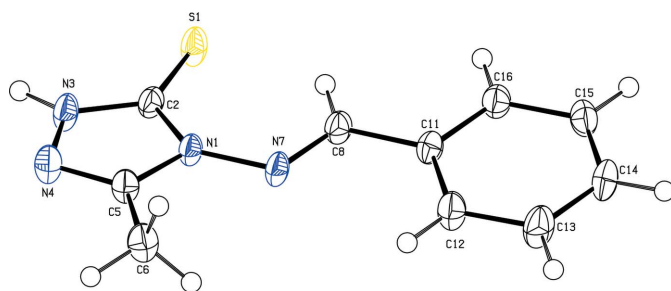


The molecular structure of the title compound, (I), is shown in Fig. 1. Bond lengths and angles can be regarded as normal (Cambridge Structural Database, Version 5.28, November 2006; *Mogul* Version 1.1; Allen, 2002; Bruno *et al.*, 2004). The acyclic C=N double bond is *trans* configured. The angle between the two rings is 10.25 (7)°.

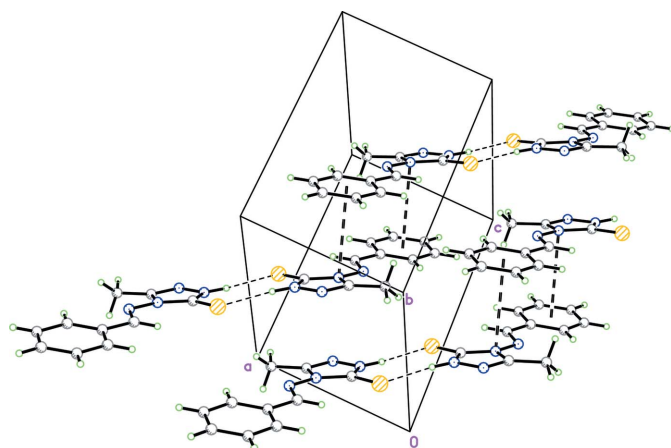
N—H···S hydrogen bonds (Table 1) link the molecules to form centrosymmetric dimers (Fig. 2). The crystal packing is further stabilized by  $\pi$ – $\pi$  interactions between atom N1 of the triazole ring and the phenyl ring of a parallel molecule at (1 – *x*, 1 – *y*, 1 – *z*), with an N1···Cg(phenyl) [Cg is the centroid] distance of 3.649 Å. The molecules crystallize in layers parallel to the (122) plane.

Received 13 February 2007

Accepted 15 February 2007



**Figure 1**  
The molecular structure of the title compound, with the atom numbering. Displacement ellipsoids are drawn at the 50% probability level.



**Figure 2**  
A packing diagram of the title compound. Hydrogen bonds are shown as dashed lines and  $\pi$ - $\pi$  interactions are shown as bold dashed lines.

## Experimental

A mixture of 4-amino-5-methyl-2,4-dihydro-3H-1,2,4-triazole-3-thione (1.3 g, 0.01 mol) and benzaldehyde (1.06 g, 0.01 mol) in absolute ethanol (25 ml) containing 2 drops of sulfuric acid, was refluxed for about 5 h. On cooling, the solid which separated was filtered off and recrystallized from ethanol (m.p. 468–470 K). Analysis (%) for  $C_{10}H_{10}N_4S$ , found (calculated): C 54.91 (55.02), H 4.48 (4.62), N 5.52 (25.67), S 14.55 (14.69).

### Crystal data

$C_{10}H_{10}N_4S$   
 $M_r = 218.28$   
 Triclinic,  $P\bar{1}$   
 $a = 6.9007$  (9) Å  
 $b = 7.3551$  (10) Å  
 $c = 11.2501$  (17) Å  
 $\alpha = 75.288$  (11)°  
 $\beta = 86.091$  (12)°  
 $\gamma = 71.424$  (10)°  
 $V = 523.46$  (13) Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.28$  mm<sup>-1</sup>  
 $T = 173$  (2) K  
 $0.48 \times 0.48 \times 0.36$  mm

### Data collection

Stoe IPDS II two-circle diffractometer  
 Absorption correction: multi-scan (MULABS; Spek, 2003; Blessing, 1995)  
 $T_{\min} = 0.878$ ,  $T_{\max} = 0.906$

5717 measured reflections  
 1957 independent reflections  
 1820 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.041$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.037$   
 $wR(F^2) = 0.102$   
 $S = 1.06$   
 1957 reflections  
 142 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.31$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.30$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

D—H...A	D—H	H...A	D...A	D—H...A
N3—H3...S1 <sup>i</sup>	0.94 (2)	2.38 (2)	3.3093 (14)	172 (2)

Symmetry code: (i)  $-x, -y + 1, -z + 2$ .

All H atoms were found in a difference map. The H atom bonded to N was refined freely. The remaining H atoms were refined using a riding model, with C—H distances ranging from 0.95 to 0.99 Å, and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  or  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$ . The methyl group was allowed to rotate but not to tip.

Data collection: X-Area (Stoe & Cie, 2001); cell refinement: X-Area; data reduction: X-Area; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003) and XP in SHELXTL-Plus (Sheldrick, 1991); software used to prepare material for publication: SHELXL97.

BKS thanks the AICTE, Government of India, for financial assistance through the Career Award for Young Teachers' Scheme. KS thanks Mangalore University for research facilities.

## References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.  
 Ambalavanan, P., Palani, K., Ponnuswamy, M. N., Thirumuruhan, R. A., Yathirajan, H. S., Prabhuswamy, B., Raju, C. R., Nagaraja, P. & Mohana, K. N. (2003). *Mol. Cryst. Liq. Cryst.* **393**, 67–73.  
 Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.  
 Bruno, I. J., Cole, J. C., Kessler, M., Luo, J., Motherwell, W. D. S., Purkis, L. H., Smith, B. R., Taylor, R., Cooper, R. I., Harris, S. E. & Orpen, A. G. (2004). *J. Chem. Inf. Comput. Sci.* **44**, 2133–2144.  
 Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
 Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.  
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.  
 Stoe & Cie (2001). *X-Area*. Stoe & Cie, Darmstadt, Germany.  
 Yathirajan, H. S., Ponnuswamy, M. N., Raju, C. R., Prabhuswamy, B., Palani, K., Nagaraja, P. & Nagaraj, B. (2005). *Anal. Sci.* **21**, x1–x2.