

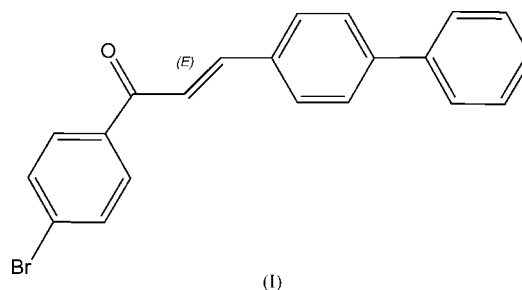
Andreas Fischer,<sup>a\*</sup> H. S.  
Yathirajan,<sup>b</sup> B. V. Ashalatha,<sup>c</sup>  
B. Narayana<sup>c</sup> and B. K. Sarojini<sup>d</sup><sup>a</sup>Inorganic Chemistry, School of Chemical Science and Engineering, Royal Institute of Technology (KTH), 100 44 Stockholm, Sweden, <sup>b</sup>Department of Studies in Chemistry, University of Mysore, Manasagangotri, Mysore 570 006, India, <sup>c</sup>Department of Chemistry, Mangalore University, Mangalagangotri 574 199, India, and <sup>d</sup>Department of Chemistry, P. A. College of Engineering, Nadupadavu, Mangalore 574 153, India

Correspondence e-mail: afischer@kth.se

**Key indicators**Single-crystal X-ray study  
 $T = 296\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.016\text{ \AA}$   
 $R$  factor = 0.064  
 $wR$  factor = 0.158  
Data-to-parameter ratio = 11.7For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.**(2E)-3-(Biphenyl-4-yl)-1-(4-bromophenyl)-prop-2-en-1-one**The title compound,  $\text{C}_{21}\text{H}_{15}\text{BrO}$ , was obtained from 4-bromoacetophenone and biphenyl-4-carbaldehyde. The geometry of the molecule is unexceptional. The compound crystallizes isostructurally with the corresponding chloro compound.

Received 30 January 2007

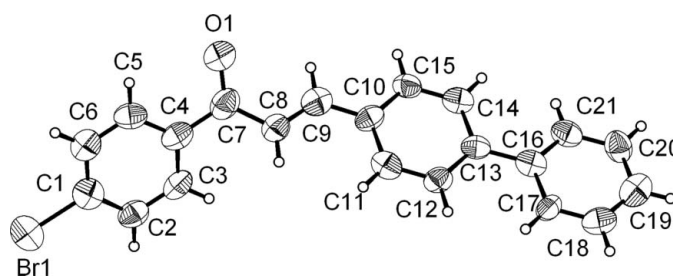
Accepted 9 February 2007

**Comment**For an introduction, see Fischer *et al.* (2007a).

The title chalcone, (I), was prepared by treating 4-bromoacetophenone with biphenyl-4-carbaldehyde in the presence of KOH.

Fig. 1 shows the molecular structure. The geometry of the molecule is unexceptional. The molecule deviates significantly from planarity [dihedral angles of  $5.4(3)^\circ$  within the biphenyl group and  $48.4(3)^\circ$  between the C10–C15 ring and the bromophenyl ring]. The compound is isostructural with the corresponding chloro compound (Fischer *et al.*, 2007b).**Experimental**

4-Bromoacetophenone (1.99 g, 0.01 mol) in methanol (20 ml) was mixed with biphenyl-4-carbaldehyde (1.82 g, 0.01 mol) and the mixture was treated with a 30% potassium hydroxide solution (3 ml) at 278 K. The reaction mixture was then brought to room tempera-

**Figure 1**

The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level.

ture and stirred for 3 h. The precipitated solid was filtered off, washed with water, dried and recrystallized from acetone (m.p. 449–451 K). Analysis (%) for C<sub>21</sub>H<sub>15</sub>BrO found (Calculated): C 69.32 (69.44), H 4.10 (4.16).

*Crystal data*

C <sub>21</sub> H <sub>15</sub> BrO	V = 1675.9 (15) Å <sup>3</sup>
M <sub>r</sub> = 363.26	Z = 4
Monoclinic, Cc	Mo Kα radiation
a = 37.35 (3) Å	μ = 2.45 mm <sup>-1</sup>
b = 7.3663 (15) Å	T = 296 K
c = 6.0978 (17) Å	0.44 × 0.23 × 0.05 mm
β = 92.63 (4)°	

*Data collection*

Bruker–Nonius KappaCCD diffractometer	4893 measured reflections
Absorption correction: numerical (Herrendorf & Bärnighausen, 1997)	2434 independent reflections
T <sub>min</sub> = 0.530, T <sub>max</sub> = 0.864	1640 reflections with I > 2σ(I)
	R <sub>int</sub> = 0.075

*Refinement*

R[F <sup>2</sup> > 2σ(F <sup>2</sup> )] = 0.064	Δρ <sub>max</sub> = 0.24 e Å <sup>-3</sup>
wR(F <sup>2</sup> ) = 0.158	Δρ <sub>min</sub> = -0.34 e Å <sup>-3</sup>
S = 1.17	Absolute structure: Flack (1983),
2434 reflections	924 Friedel pairs
208 parameters	Flack parameter: -0.01 (3)
H-atom parameters constrained	

H atoms were placed at calculated positions and refined as riding on the respective carrier atom, with C–H = 0.93 Å and U<sub>iso</sub>(H) = 1.2U<sub>eq</sub>(C). The structure appears to exhibit turbostratic disorder, which could be detected in precession photographs that were simu-

lated from the CCD data. The disorder was accounted for in the data processing with *EVALCCD* (Duisenberg *et al.*, 2003).

Data collection: *COLLECT* (Nonius, 1999); cell refinement: *DIRAX/LSQ* (Duisenberg, 1992); data reduction: *EVALCCD* (Duisenberg *et al.*, 2003); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 2006); software used to prepare material for publication: *publCIF* (Westrip, 2007).

One of the authors (BKS) thanks AICTE, Government of India, for financial assistance through the Career Award for Young Teacher's Scheme, and BVA thanks Mangalore University for permission to carry out the research work. The Swedish Research Council (VR) is acknowledged for providing funding for the single-crystal diffractometer.

**References**

Brandenburg, K. (2006). *DIAMOND*. Release 3.1d. Crystal Impact GbR, Bonn, Germany.

Duisenberg, A. J. M. (1992). *J. Appl. Cryst.* **25**, 92–96.

Duisenberg, A. J. M., Kroon-Batenburg, L. M. J. & Schreurs, A. M. M. (2003). *J. Appl. Cryst.* **36**, 220–229.

Fischer, A., Yathirajan, H. S., Ashalatha, B. V., Narayana, B. & Sarojini, B. K. (2007b). *Acta Cryst.* **E63**, o1353–o1354.

Fischer, A., Yathirajan, H. S., Ashalatha, B. V., Narayana, B. & Sarojini, B. K. (2007a). *Acta Cryst.* **E63**, o1349–o1350.

Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.

Herrendorf, W. & Bärnighausen, H. (1997). *HABITUS*. University of Karlsruhe, Germany.

Nonius (1999). *COLLECT*. Nonius BV, Delft, The Netherlands.

Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.

Westrip, S. P. (2007). *publCIF*. In preparation.