

Ethopropazinium picrate

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Key indicators

Single-crystal X-ray study
 $T = 173$ K
Mean $\sigma(\text{C}-\text{C}) = 0.005$ Å
 R factor = 0.077
 wR factor = 0.212
Data-to-parameter ratio = 12.7

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

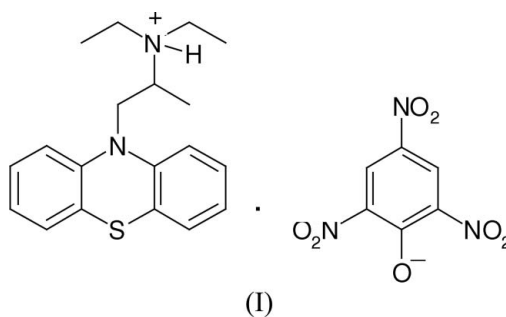
The title compound [systematic name: 10-[2-(diethylamino)propyl]phenothiazinium 2,4,6-trinitrophenolate], $\text{C}_{19}\text{H}_{25}\text{N}_2\text{S}^+\cdot\text{C}_6\text{H}_2\text{N}_3\text{O}_7^-$, is a pharmacologically active compound. The dihedral angle between the two outer aromatic rings of the phenothiazine unit is $38.64(12)^\circ$. The crystal packing is stabilized by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds and several weak $\text{C}-\text{H}\cdots\text{O}$ contacts. The molecular conformation of the cation does not change significantly when it is crystallized with chloride or perrhenate as the anion.

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Comment

Ethopropazine is an anticholinergic agent with some antihistaminic and ganglionic blocking activity (Bratfos & Haug, 1979). The present work results in the formation of a salt by the interaction between ethopropazinium hydrochloride and 2,4,6-trinitrophenol in an aqueous medium.



A perspective view of the title compound, (I), is shown in Fig. 1. Bond lengths and angles can be regarded as normal (Cambridge Structural Database, Version 5.28, November 2006; *MOGUL*, Version 1.1; Allen, 2002; Bruno *et al.*, 2004). The dihedral angle between the two aromatic rings of the phenothiazine unit is $38.64(12)^\circ$. Crystallographic data for ethopropazinium chloride, (Ia) (Marsau & Calas, 1971; Klein *et al.*, 1994), and ethopropazinium perrhenate, (Ib) (Gowda *et al.*, 1994), have been published, but since there is a coordinate error in the report by Marsau & Calas (1971), the structure given by Klein *et al.* (1994) is employed for comparison. Least-squares overlays of the ethopropazinium cations of (I) and (Ia) (r.m.s. deviation 0.066 Å) as well as of (I) and (Ib) (r.m.s. deviation 0.120 Å), fitting only the phenothiazine units, are shown in Figs. 2 and 3. As can be seen, the molecular conformation is not significantly different in the compared structures. The crystal packing is stabilized by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds and several weak $\text{C}-\text{H}\cdots\text{O}$ contacts (Table 1).

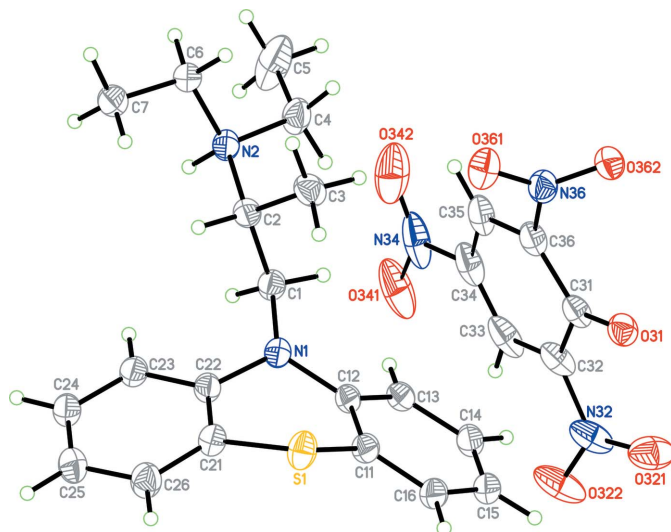


Figure 1
The structure of the title compound, with the atom numbering. Displacement ellipsoids are drawn at the 30% probability level.

Experimental

Profenamine hydrochloride (0.7010 g, 0.02 *M*) and picric acid (0.4610 g, 0.02 *M*) were dissolved in distilled water (100 ml), mixed and stirred in a beaker at room temperature. The separated yellow salt was washed well with distilled water, filtered and dried in a vacuum desiccator over phosphorus pentoxide. The complex was recrystallized from acetonitrile (m.p. 378 K).

Crystal data

$C_{19}H_{25}N_2S^+ \cdot C_6H_2N_3O_7^-$	$V = 5014.3 (6) \text{ \AA}^3$
$M_r = 541.58$	$Z = 8$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation
$a = 36.876 (3) \text{ \AA}$	$\mu = 0.19 \text{ mm}^{-1}$
$b = 8.4622 (4) \text{ \AA}$	$T = 173 (2) \text{ K}$
$c = 16.5727 (11) \text{ \AA}$	$0.29 \times 0.26 \times 0.25 \text{ mm}$
$\beta = 104.163 (6)^\circ$	

Data collection

Stoe IPDS-II two-circle diffractometer	22001 measured reflections
Absorption correction: multi-scan (<i>MULABS</i> ; Spek, 2003; Blessing, 1995)	4421 independent reflections
$T_{\min} = 0.938$, $T_{\max} = 0.945$	3505 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.081$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.077$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.212$	$\Delta\rho_{\max} = 0.46 \text{ e \AA}^{-3}$
$S = 1.06$	$\Delta\rho_{\min} = -0.55 \text{ e \AA}^{-3}$
4421 reflections	
347 parameters	

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N2-H2\cdots O31^i$	0.95 (5)	1.92 (5)	2.841 (4)	161 (4)
$C1-H1A\cdots O31^i$	0.99	2.45	3.234 (4)	136
$C23-H23\cdots O31^i$	0.95	2.37	3.316 (4)	175

Symmetry code: (i) $x, -y + 1, z - \frac{1}{2}$.

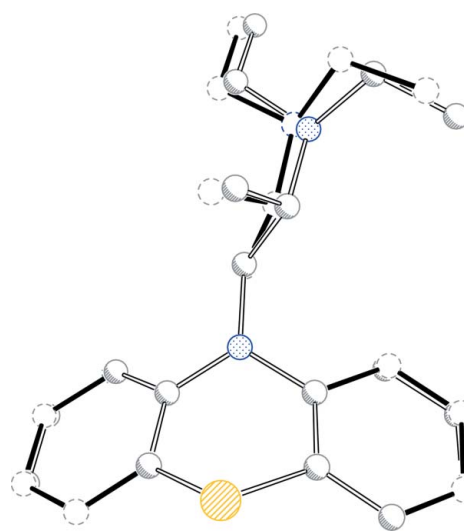


Figure 2
Least-squares fit of the ethopropazinium cations in (I) (full bonds) and (Ia) (open bonds). H atoms have been omitted.



Figure 3
Least-squares fit of the ethopropazinium cations in (I) (full bonds) and (Ib) (open bonds). H atoms have been omitted.

H atoms were found in a difference map, but the C-bound H atoms were refined using a riding model, with C—H ranging from 0.95 to 0.99 \AA and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$. The H atom bonded to N was freely refined.

Data collection: *X-Area* (Stoe & Cie, 2001); cell refinement: *X-Area*; data reduction: *X-Area*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* in *SHELXTL-Plus* (Sheldrick, 1991); software used to prepare material for publication: *SHELXL97*.

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References

- Allen, F. H. (2002). *Acta Cryst.* **B58**, 380–388.
- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.
- Bratfos, O. & Haug, J. O. (1979). *Acta Psychiatr. Scand.* **60**, 1–9.

- Bruno, I. J., Cole, J. C., Kessler, M., Luo, J., Motherwell, W. D. S., Purkis, L. H., Smith, B. R., Taylor, R., Cooper, R. I., Harris, S. E. & Orpen, A. G. (2004). *J. Chem. Inf. Comput. Sci.* **44**, 2133–2144.
- Gowda, N. M. M., Zhang, L. & Barnes, C. L. (1994). *J. Chem. Crystallogr.* **24**, 89–93.
- Klein, C. L., Lear, J., O'Rourke, S., Williams, S. & Liang, L. (1994). *J. Pharm. Sci.* **83**, 1253–1256.
- Marsau, P. & Calas, M.-R. (1971). *Acta Cryst.* **B27**, 2058–2062.
- Sheldrick, G. M. (1991). *SHELXTL-Plus*. Release 4.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Stoe & Cie (2001). *X-Area*. Stoe & Cie, Darmstadt, Germany.