

N-(2-Nitrobenzylidene)aniline

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Key indicators

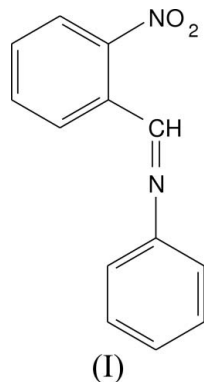
Single-crystal X-ray study
 $T = 295$ K
Mean $\sigma(\text{C}-\text{C}) = 0.003$ Å
 R factor = 0.051
 wR factor = 0.172
Data-to-parameter ratio = 12.4

For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

In the title compound, $\text{C}_{13}\text{H}_{10}\text{N}_2\text{O}_2$, the dihedral angle between the two six-membered aromatic rings is $60.49(9)^\circ$. The molecule adopts an *E* configuration with respect to the imine $\text{C}=\text{N}$ bond.

Comment

Imines are synthesized by the reaction of an aldehyde or a ketone with a primary amine (Patai, 1970). In order to obtain the desired imine, the by-product, water, must be removed in order to drive the equilibrium in favour of the imine (Layer, 1963). This result can be achieved by the use of a Dean–Stark trap or molecular sieves (Bose *et al.*, 1967). As part of our work in the area of catalytic transfer hydrogenation, we required a number of imines as starting materials. With this background, the title compound, (I), was synthesized and we report its crystal structure here.



A perspective view of (I) is shown in Fig. 1. In compound (I), the dihedral angle between the two six-membered aromatic rings is $60.49(9)^\circ$. As (I) crystallizes in a centrosymmetric space group, $P2_1/c$, there is no spontaneous resolution. The $\text{C}12-\text{N}11-\text{C}10-\text{C}3$ torsion angle of $175.05(14)^\circ$ indicates that the molecule adopts an *E* configuration with respect to the imine $\text{C}=\text{N}$ bond. This is comparable with the corresponding value of $176.2(3)^\circ$ reported earlier (Akitsu *et al.*, 2004). The nitro group deviates from the plane of the phenyl ring, as indicated by the $\text{O}9-\text{N}7-\text{C}4-\text{C}3$ and $\text{O}8-\text{N}7-\text{C}4-\text{C}5$ torsion angles of $-35.0(2)^\circ$ and $-36.2(2)^\circ$, respectively.

Experimental

To a solution of *o*-nitrobenzaldehyde (0.75 g, 5 mmol) in dichloromethane (6 ml), aniline (0.45 ml, 5 mmol) was slowly added dropwise with constant stirring at 273 K in the presence of molecular sieves.

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The reaction mixture was further stirred for a further 12 h at room temperature. The molecular sieves were removed by filtration and the filtrate was washed with KHSO_4 . The solvent was removed under reduced pressure and the resulting product was recrystallized using methanol.

Crystal data

$\text{C}_{13}\text{H}_{10}\text{N}_2\text{O}_2$
 $M_r = 226.23$
 Monoclinic, $P2_1/c$
 $a = 11.178$ (13) Å
 $b = 7.718$ (5) Å
 $c = 13.217$ (16) Å
 $\beta = 94.064$ (3)°
 $V = 1137$ (2) Å³

$Z = 4$
 $D_x = 1.321$ Mg m⁻³
 Mo $K\alpha$ radiation
 $\mu = 0.09$ mm⁻¹
 $T = 295$ (2) K
 Block, pale yellow
 $0.25 \times 0.23 \times 0.22$ mm

Data collection

MacScience DIPLabo 32001
 diffractometer
 ω scans
 Absorption correction: none
 3583 measured reflections

1923 independent reflections
 1555 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.020$
 $\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.051$
 $wR(F^2) = 0.172$
 $S = 1.00$
 1923 reflections
 155 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.1279P)^2 + 0.0772P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.27$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.16$ e Å⁻³
 Extinction correction: *SHELXL97*
 (Sheldrick, 1997)
 Extinction coefficient: 0.081 (12)

H atoms were placed in idealized positions and allowed to ride on their parent atoms, with C–H = 0.93 Å and $U_{\text{iso}}(\text{H}) 1.2U_{\text{eq}}(\text{carrier atom})$.

Data collection: *XPRESS* (MacScience, 2002); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO* (Otwinowski & Minor, 1997) and *SCALEPACK*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997);

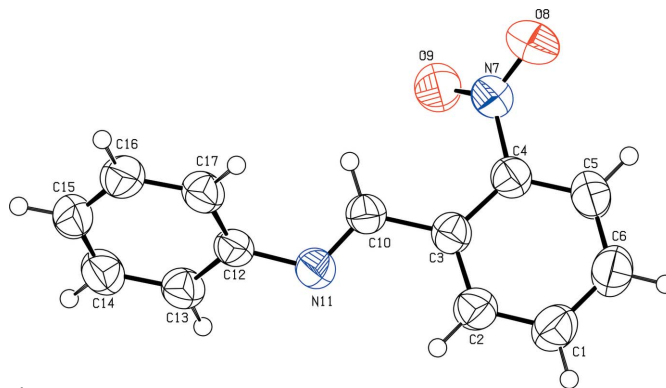


Figure 1
 A view of (I), with 50% probability displacement ellipsoids.

program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003) and *ORTEPII* (Johnson, 1976); software used to prepare material for publication: *PLATON*.

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