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## Structure Reports

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## (2E)-1-(3-Bromophenyl)-3-(6-methoxy-2-naphthyl)prop-2-en-1-one

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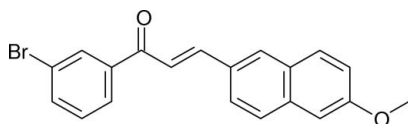
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Key indicators: single-crystal X-ray study;  $T = 120$  K; mean  $\sigma(\text{C}-\text{C}) = 0.010$  Å;  $R$  factor = 0.082;  $wR$  factor = 0.163; data-to-parameter ratio = 17.0.

In the title compound,  $\text{C}_{20}\text{H}_{15}\text{BrO}_2$ , the prop-2-en-1-one fragment is substantially twisted [ $\text{C}-\text{C}-\text{O} = 23.0$  ( $11^\circ$ )]. The dihedral angle between the benzene and naphthalene rings is  $44.28$  ( $13^\circ$ ). The only possible directional interactions in the crystal are weak  $\text{C}-\text{H}\cdots\pi$  contacts, which generate (001) sheets.

## Related literature

For related structures, see: Yathirajan *et al.* (2007*a,b*); Jasinski *et al.* (2009). For background to the non-linear optical properties of chalcones, see: Sarojini *et al.* (2006). For reference structural data, see: Allen *et al.* (1987).



## Experimental

## Crystal data

$\text{C}_{20}\text{H}_{15}\text{BrO}_2$   
 $M_r = 367.23$   
Orthorhombic,  $Pbca$   
 $a = 14.0955$  (14) Å  
 $b = 6.1295$  (6) Å  
 $c = 36.119$  (4) Å

$V = 3120.6$  (5) Å<sup>3</sup>  
 $Z = 8$   
Mo  $K\alpha$  radiation  
 $\mu = 2.64$  mm<sup>-1</sup>  
 $T = 120$  K  
 $0.11 \times 0.09 \times 0.03$  mm

## Data collection

Nonius KappaCCD diffractometer  
Absorption correction: multi-scan  
(*SADABS*; Bruker, 2003)  
 $T_{\min} = 0.760$ ,  $T_{\max} = 0.925$   
28579 measured reflections  
3545 independent reflections  
1719 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.228$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.082$   
 $wR(F^2) = 0.163$   
 $S = 1.05$   
3545 reflections  
209 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.61$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.63$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

Cg2 is the centroid of the C3–C8 ring.

| $D-H\cdots A$                                     | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|---|-------|-------------|-------------|---------------|
| $\text{C4}-\text{H4}\cdots\text{Cg2}^{\text{i}}$  | 0.95  | 2.70        | 3.432 (6)   | 134           |
| $\text{C7}-\text{H7}\cdots\text{Cg2}^{\text{ii}}$ | 0.95  | 2.80        | 3.520 (6)   | 134           |

Symmetry codes: (i)  $-x + \frac{1}{2}, y - \frac{3}{2}, z$ ; (ii)  $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$ .

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO* (Otwinowski & Minor, 1997), *SCALEPACK* and *SORTAV* (Blessing, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK2699).

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Yathirajan, H. S., Mayekar, A. N., Sarojini, B. K., Narayana, B. & Bolte, M. (2007*b*). *Acta Cryst.* **E63**, o1012–o1013.

## supporting information

*Acta Cryst.* (2010). E66, o2552 [doi:10.1107/S1600536810035117]

**(2E)-1-(3-Bromophenyl)-3-(6-methoxy-2-naphthyl)prop-2-en-1-one**

William T. A. Harrison, A. N. Mayekar, H. S. Yathirajan and B. Narayana

**S1. Comment**

The title compound, (I), (Fig. 1), was prepared as part of our ongoing studies (Yathirajan *et al.*, 2007*a,b*; Jasinski *et al.*, 2009) of substituted phenyl/naphthyl chalcone derivatives as possible candidates for non-linear optical materials (Sarojini *et al.*, 2006). However, (I) crystallizes in a centrosymmetric space group, thus its second-harmonic generation (SHG) response must be zero.

The prop-2-en-1-one (enone) fragment in (I) is substantially twisted, as indicated by the C11—C12—C13—O2 torsion angle of 23.0 (11)°. The dihedral angle between the aromatic ring systems is 44.28 (13)°. Equivalent data for related structures are as follows: (2E)-1-(2,4-dichlorophenyl)-3-(6-methoxy-2-naphthyl)prop-2-en-1-one (Yathirajan *et al.*, 2007*a*): -10.9 (2) and 44.94 (4)°; (2E)-3-(6-methoxy-2-naphthyl)-1-phenylprop-2-en-1-one (Yathirajan *et al.*, 2007*b*): -15.9 (4) and 14.9 (8)°; (2E)-1-(2-hydroxyphenyl)-3-(6-methoxy-2-naphthyl)prop-2-en-1-one (Jasinski *et al.*, 2009): -14.9 (2) and 31.7 (3)°. Otherwise, the bond lengths for (I) fall within their expected ranges (Allen *et al.*, 1987).

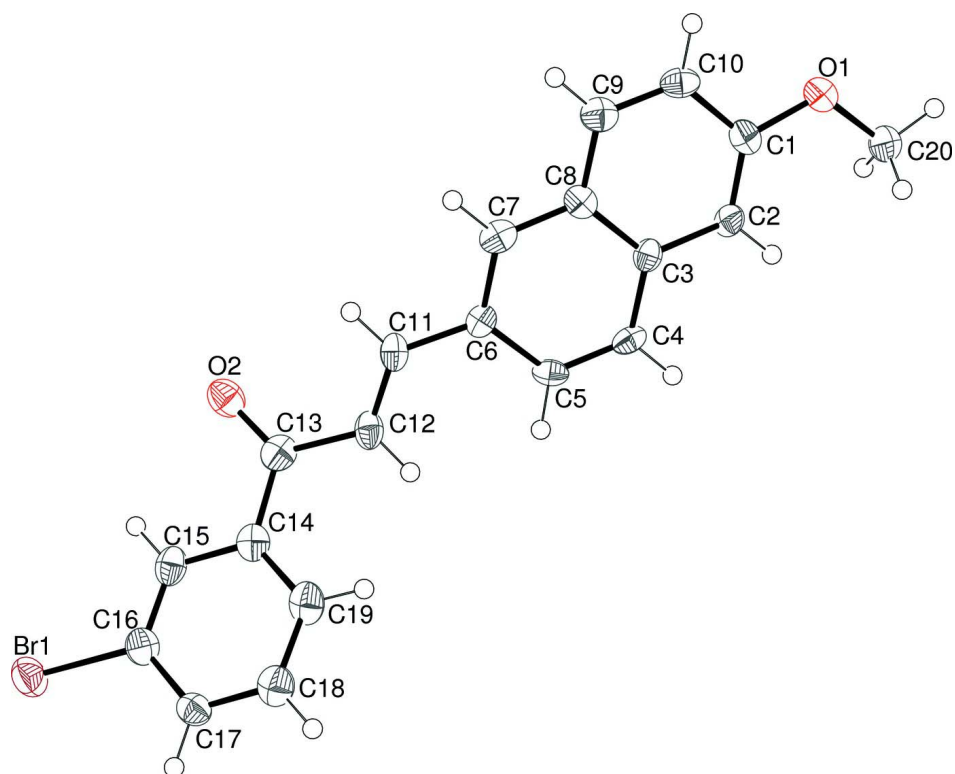
In the crystal of (I), the only possible directional interactions between molecules are weak C—H $\cdots$  $\pi$  contacts in which the C3—C8 ring of the naphthyl moiety provides both the C—H donor groups and the aromatic acceptor surface (Table 1, Fig. 2). Together, these generate (001) sheets.

**S2. Experimental**

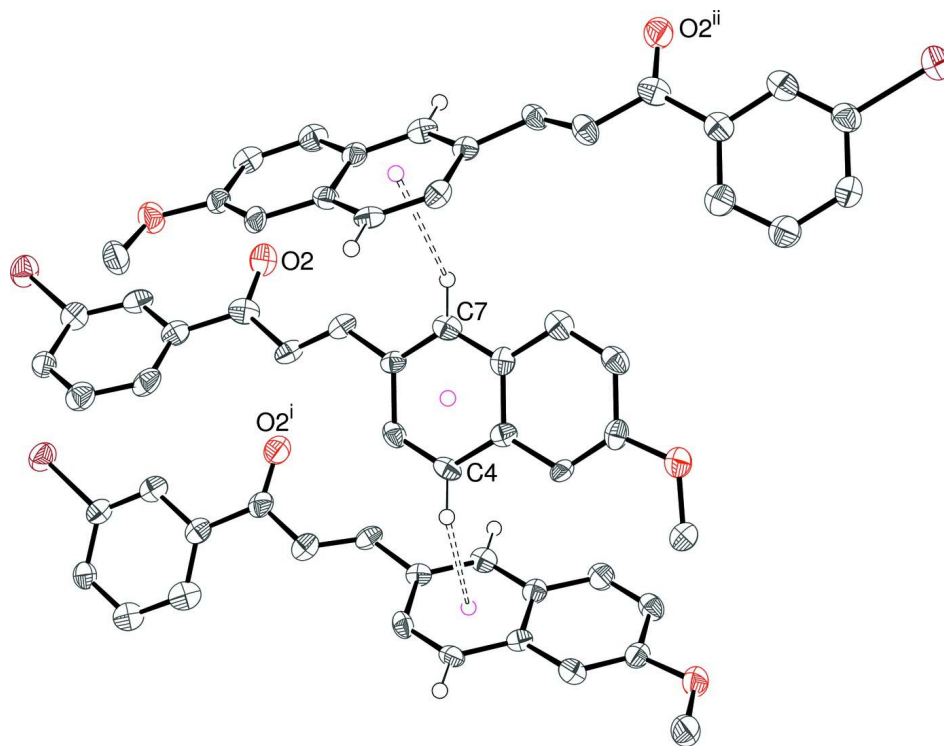
To a thoroughly stirred solution of 6-methoxy-2-naphthaldehyde (1.86 g, 0.01 mol) and 3-bromoacetophenone (1.99 g, 0.01 mol) in 25 ml methanol, 5 ml of 40% KOH solution was added. The reaction mixture was stirred overnight and the solid separated was collected by filtration. The product obtained was recrystallized from methanol. Colourless slabs of (I) were grown by the slow evaporation of the ethylacetate solution (m.p. 427–429 K).

**S3. Refinement**

The crystal studied was a weak scatterer, which may correlate with the high  $R_{\text{int}}$  value. The hydrogen atoms were geometrically placed (C—H = 0.95–0.98 Å) and refined as riding with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  or  $1.5U_{\text{eq}}(\text{methyl C})$ . A rotating rigid-group model was applied to the methyl group.

**Figure 1**

View of the molecular structure of (I) showing 50% displacement ellipsoids (arbitrary spheres for the H atoms).



**Figure 2**

Partial packing diagram for (I) showing the possible weak C—H... $\pi$  contacts. All H atoms except H4 and H7 omitted for clarity. Symmetry codes: (i)  $1/2-x, y-1/2, z$ ; (ii)  $1-x, 1/2+y, 1/2-z$ .

**(2E)-1-(3-Bromophenyl)-3-(6-methoxy-2-naphthyl)prop-2-en-1-one**

*Crystal data*

$C_{20}H_{15}BrO_2$

$M_r = 367.23$

Orthorhombic, *Pbca*

Hall symbol: -P 2ac 2ab

$a = 14.0955$  (14) Å

$b = 6.1295$  (6) Å

$c = 36.119$  (4) Å

$V = 3120.6$  (5) Å<sup>3</sup>

$Z = 8$

$F(000) = 1488$

$D_x = 1.563$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 55128 reflections

$\theta = 2.9$ – $27.5^\circ$

$\mu = 2.64$  mm<sup>-1</sup>

$T = 120$  K

Slab, colourless

$0.11 \times 0.09 \times 0.03$  mm

*Data collection*

Nonius KappaCCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\omega$  and  $\phi$  scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2003)

$T_{\min} = 0.760$ ,  $T_{\max} = 0.925$

28579 measured reflections

3545 independent reflections

1719 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.228$

$\theta_{\max} = 27.5^\circ$ ,  $\theta_{\min} = 3.1^\circ$

$h = -18 \rightarrow 18$

$k = -7 \rightarrow 7$

$l = -46 \rightarrow 46$

Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.082$   
 $wR(F^2) = 0.163$   
 $S = 1.05$   
 3545 reflections  
 209 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0343P)^2 + 12.2862P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.61 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.63 \text{ e } \text{\AA}^{-3}$

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

|     | $x$        | $y$         | $z$          | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|------------|-------------|--------------|----------------------------------|
| C1  | 0.3881 (4) | 0.3963 (12) | 0.14070 (18) | 0.0269 (16)                      |
| C2  | 0.3536 (5) | 0.2944 (11) | 0.17160 (18) | 0.0260 (16)                      |
| H2  | 0.3224     | 0.1576      | 0.1695       | 0.031*                           |
| C3  | 0.3647 (4) | 0.3940 (11) | 0.20683 (18) | 0.0244 (15)                      |
| C4  | 0.3332 (4) | 0.2939 (11) | 0.23966 (18) | 0.0247 (16)                      |
| H4  | 0.3057     | 0.1526      | 0.2384       | 0.030*                           |
| C5  | 0.3411 (4) | 0.3933 (11) | 0.27329 (17) | 0.0257 (15)                      |
| H5  | 0.3192     | 0.3202      | 0.2949       | 0.031*                           |
| C6  | 0.3820 (4) | 0.6074 (12) | 0.27654 (18) | 0.0226 (15)                      |
| C7  | 0.4144 (4) | 0.7064 (11) | 0.24441 (19) | 0.0266 (17)                      |
| H7  | 0.4418     | 0.8477      | 0.2460       | 0.032*                           |
| C8  | 0.4083 (4) | 0.6051 (12) | 0.20954 (18) | 0.0249 (16)                      |
| C9  | 0.4425 (5) | 0.7045 (12) | 0.1768 (2)   | 0.0316 (18)                      |
| H9  | 0.4724     | 0.8432      | 0.1782       | 0.038*                           |
| C10 | 0.4332 (4) | 0.6034 (12) | 0.14329 (19) | 0.0290 (16)                      |
| H10 | 0.4569     | 0.6718      | 0.1216       | 0.035*                           |
| C11 | 0.3820 (4) | 0.7262 (12) | 0.31156 (18) | 0.0262 (18)                      |
| H11 | 0.4035     | 0.8730      | 0.3107       | 0.031*                           |
| C12 | 0.3550 (5) | 0.6518 (11) | 0.34459 (18) | 0.0290 (17)                      |
| H12 | 0.3356     | 0.5039      | 0.3468       | 0.035*                           |
| C13 | 0.3544 (5) | 0.7926 (12) | 0.3780 (2)   | 0.0305 (17)                      |
| C14 | 0.3642 (4) | 0.6878 (12) | 0.4150 (2)   | 0.0276 (17)                      |
| C15 | 0.3486 (5) | 0.8158 (12) | 0.44671 (19) | 0.0312 (17)                      |
| H15 | 0.3274     | 0.9622      | 0.4442       | 0.037*                           |
| C16 | 0.3638 (5) | 0.7308 (12) | 0.4810 (2)   | 0.0313 (17)                      |

|      |             |              |              |             |
|------|-------------|--------------|--------------|-------------|
| C17  | 0.3927 (5)  | 0.5156 (12)  | 0.48559 (19) | 0.0290 (18) |
| H17  | 0.4014      | 0.4555       | 0.5096       | 0.035*      |
| C18  | 0.4085 (5)  | 0.3918 (13)  | 0.4543 (2)   | 0.0363 (18) |
| H18  | 0.4305      | 0.2461       | 0.4571       | 0.044*      |
| C19  | 0.3934 (5)  | 0.4728 (12)  | 0.4191 (2)   | 0.035 (2)   |
| H19  | 0.4029      | 0.3827       | 0.3981       | 0.042*      |
| C20  | 0.3351 (5)  | 0.1184 (12)  | 0.10028 (19) | 0.0400 (19) |
| H20A | 0.3318      | 0.0822       | 0.0739       | 0.060*      |
| H20B | 0.2707      | 0.1298       | 0.1103       | 0.060*      |
| H20C | 0.3697      | 0.0036       | 0.1135       | 0.060*      |
| O1   | 0.3827 (3)  | 0.3202 (8)   | 0.10487 (13) | 0.0351 (13) |
| O2   | 0.3495 (4)  | 0.9934 (9)   | 0.37483 (13) | 0.0361 (12) |
| Br1  | 0.34948 (6) | 0.90763 (13) | 0.52413 (2)  | 0.0403 (3)  |

*Atomic displacement parameters (Å<sup>2</sup>)*

|     | $U^{11}$   | $U^{22}$   | $U^{33}$   | $U^{12}$   | $U^{13}$   | $U^{23}$    |
|-----|------------|------------|------------|------------|------------|-------------|
| C1  | 0.024 (4)  | 0.033 (4)  | 0.024 (4)  | 0.007 (4)  | -0.003 (3) | -0.002 (4)  |
| C2  | 0.027 (4)  | 0.023 (4)  | 0.027 (4)  | -0.005 (4) | -0.011 (4) | -0.001 (3)  |
| C3  | 0.022 (4)  | 0.022 (4)  | 0.029 (4)  | 0.002 (4)  | 0.000 (3)  | -0.003 (4)  |
| C4  | 0.021 (4)  | 0.019 (4)  | 0.034 (4)  | 0.002 (3)  | 0.004 (3)  | 0.006 (3)   |
| C5  | 0.022 (3)  | 0.031 (4)  | 0.024 (4)  | -0.002 (4) | 0.003 (3)  | 0.008 (4)   |
| C6  | 0.014 (3)  | 0.023 (4)  | 0.030 (4)  | -0.001 (3) | -0.002 (3) | -0.001 (4)  |
| C7  | 0.013 (3)  | 0.029 (4)  | 0.038 (5)  | -0.001 (3) | 0.002 (3)  | 0.004 (4)   |
| C8  | 0.018 (3)  | 0.031 (4)  | 0.026 (4)  | 0.005 (4)  | -0.001 (3) | 0.000 (4)   |
| C9  | 0.027 (4)  | 0.034 (4)  | 0.034 (5)  | 0.002 (4)  | 0.003 (4)  | 0.003 (4)   |
| C10 | 0.022 (4)  | 0.032 (4)  | 0.033 (4)  | 0.000 (4)  | 0.003 (3)  | 0.009 (4)   |
| C11 | 0.019 (4)  | 0.028 (4)  | 0.031 (4)  | 0.001 (3)  | 0.000 (3)  | -0.009 (4)  |
| C12 | 0.024 (4)  | 0.031 (4)  | 0.032 (4)  | -0.006 (4) | 0.000 (4)  | -0.009 (3)  |
| C13 | 0.021 (4)  | 0.032 (5)  | 0.038 (4)  | -0.001 (4) | 0.011 (4)  | -0.001 (4)  |
| C14 | 0.016 (4)  | 0.033 (4)  | 0.034 (4)  | -0.002 (3) | 0.002 (3)  | -0.004 (4)  |
| C15 | 0.025 (4)  | 0.030 (4)  | 0.039 (4)  | -0.004 (4) | 0.002 (4)  | -0.007 (4)  |
| C16 | 0.028 (4)  | 0.034 (4)  | 0.032 (4)  | 0.003 (3)  | 0.003 (4)  | -0.004 (4)  |
| C17 | 0.027 (4)  | 0.034 (4)  | 0.026 (4)  | -0.003 (3) | -0.001 (3) | 0.002 (3)   |
| C18 | 0.030 (4)  | 0.038 (5)  | 0.041 (5)  | 0.000 (4)  | -0.005 (4) | -0.003 (5)  |
| C19 | 0.023 (4)  | 0.040 (5)  | 0.042 (5)  | -0.001 (3) | -0.011 (4) | -0.007 (4)  |
| C20 | 0.059 (5)  | 0.033 (4)  | 0.029 (4)  | -0.007 (4) | -0.005 (4) | -0.004 (4)  |
| O1  | 0.045 (3)  | 0.035 (3)  | 0.026 (3)  | -0.002 (2) | 0.001 (2)  | -0.001 (3)  |
| O2  | 0.034 (3)  | 0.044 (3)  | 0.031 (3)  | 0.006 (3)  | 0.001 (3)  | -0.002 (3)  |
| Br1 | 0.0471 (4) | 0.0440 (5) | 0.0298 (4) | 0.0048 (5) | 0.0016 (4) | -0.0062 (4) |

*Geometric parameters (Å, °)*

|        |           |         |            |
|--------|-----------|---------|------------|
| C1—C2  | 1.369 (9) | C11—H11 | 0.9500     |
| C1—O1  | 1.378 (8) | C12—C13 | 1.483 (9)  |
| C1—C10 | 1.422 (9) | C12—H12 | 0.9500     |
| C2—C3  | 1.420 (9) | C13—O2  | 1.238 (8)  |
| C2—H2  | 0.9500    | C13—C14 | 1.489 (10) |

|             |           |               |            |
|-------------|-----------|---------------|------------|
| C3—C4       | 1.407 (9) | C14—C19       | 1.389 (9)  |
| C3—C8       | 1.436 (9) | C14—C15       | 1.406 (9)  |
| C4—C5       | 1.364 (9) | C15—C16       | 1.361 (9)  |
| C4—H4       | 0.9500    | C15—H15       | 0.9500     |
| C5—C6       | 1.439 (9) | C16—C17       | 1.391 (10) |
| C5—H5       | 0.9500    | C16—Br1       | 1.909 (7)  |
| C6—C7       | 1.387 (9) | C17—C18       | 1.379 (9)  |
| C6—C11      | 1.459 (9) | C17—H17       | 0.9500     |
| C7—C8       | 1.407 (9) | C18—C19       | 1.380 (10) |
| C7—H7       | 0.9500    | C18—H18       | 0.9500     |
| C8—C9       | 1.414 (9) | C19—H19       | 0.9500     |
| C9—C10      | 1.366 (9) | C20—O1        | 1.418 (8)  |
| C9—H9       | 0.9500    | C20—H20A      | 0.9800     |
| C10—H10     | 0.9500    | C20—H20B      | 0.9800     |
| C11—C12     | 1.333 (9) | C20—H20C      | 0.9800     |
| C2—C1—O1    | 126.3 (6) | C6—C11—H11    | 116.4      |
| C2—C1—C10   | 120.8 (6) | C11—C12—C13   | 122.0 (7)  |
| O1—C1—C10   | 112.9 (6) | C11—C12—H12   | 119.0      |
| C1—C2—C3    | 119.7 (6) | C13—C12—H12   | 119.0      |
| C1—C2—H2    | 120.1     | O2—C13—C12    | 120.3 (7)  |
| C3—C2—H2    | 120.1     | O2—C13—C14    | 121.1 (7)  |
| C4—C3—C2    | 122.2 (6) | C12—C13—C14   | 118.6 (6)  |
| C4—C3—C8    | 118.1 (6) | C19—C14—C15   | 119.2 (7)  |
| C2—C3—C8    | 119.7 (6) | C19—C14—C13   | 122.2 (7)  |
| C5—C4—C3    | 122.0 (6) | C15—C14—C13   | 118.5 (6)  |
| C5—C4—H4    | 119.0     | C16—C15—C14   | 120.2 (7)  |
| C3—C4—H4    | 119.0     | C16—C15—H15   | 119.9      |
| C4—C5—C6    | 120.9 (6) | C14—C15—H15   | 119.9      |
| C4—C5—H5    | 119.6     | C15—C16—C17   | 121.2 (7)  |
| C6—C5—H5    | 119.6     | C15—C16—Br1   | 120.6 (5)  |
| C7—C6—C5    | 117.6 (6) | C17—C16—Br1   | 118.2 (5)  |
| C7—C6—C11   | 120.5 (6) | C18—C17—C16   | 118.1 (7)  |
| C5—C6—C11   | 121.7 (6) | C18—C17—H17   | 120.9      |
| C6—C7—C8    | 122.4 (6) | C16—C17—H17   | 120.9      |
| C6—C7—H7    | 118.8     | C17—C18—C19   | 122.1 (7)  |
| C8—C7—H7    | 118.8     | C17—C18—H18   | 119.0      |
| C7—C8—C9    | 122.5 (7) | C19—C18—H18   | 119.0      |
| C7—C8—C3    | 119.0 (6) | C18—C19—C14   | 119.1 (8)  |
| C9—C8—C3    | 118.5 (6) | C18—C19—H19   | 120.4      |
| C10—C9—C8   | 120.8 (7) | C14—C19—H19   | 120.4      |
| C10—C9—H9   | 119.6     | O1—C20—H20A   | 109.5      |
| C8—C9—H9    | 119.6     | O1—C20—H20B   | 109.5      |
| C9—C10—C1   | 120.4 (7) | H20A—C20—H20B | 109.5      |
| C9—C10—H10  | 119.8     | O1—C20—H20C   | 109.5      |
| C1—C10—H10  | 119.8     | H20A—C20—H20C | 109.5      |
| C12—C11—C6  | 127.2 (7) | H20B—C20—H20C | 109.5      |
| C12—C11—H11 | 116.4     | C1—O1—C20     | 115.6 (6)  |

|              |            |                 |            |
|--------------|------------|-----------------|------------|
| O1—C1—C2—C3  | -180.0 (6) | C7—C6—C11—C12   | -178.5 (7) |
| C10—C1—C2—C3 | -1.4 (9)   | C5—C6—C11—C12   | 7.2 (10)   |
| C1—C2—C3—C4  | -178.0 (6) | C6—C11—C12—C13  | -177.3 (6) |
| C1—C2—C3—C8  | 2.7 (9)    | C11—C12—C13—O2  | 23.0 (11)  |
| C2—C3—C4—C5  | -177.7 (6) | C11—C12—C13—C14 | -154.4 (6) |
| C8—C3—C4—C5  | 1.6 (9)    | O2—C13—C14—C19  | -163.5 (7) |
| C3—C4—C5—C6  | 0.2 (10)   | C12—C13—C14—C19 | 13.8 (10)  |
| C4—C5—C6—C7  | -1.1 (9)   | O2—C13—C14—C15  | 12.6 (10)  |
| C4—C5—C6—C11 | 173.4 (6)  | C12—C13—C14—C15 | -170.1 (6) |
| C5—C6—C7—C8  | 0.1 (9)    | C19—C14—C15—C16 | 1.2 (10)   |
| C11—C6—C7—C8 | -174.5 (6) | C13—C14—C15—C16 | -175.0 (6) |
| C6—C7—C8—C9  | -179.1 (6) | C14—C15—C16—C17 | -1.5 (10)  |
| C6—C7—C8—C3  | 1.7 (10)   | C14—C15—C16—Br1 | 176.6 (5)  |
| C4—C3—C8—C7  | -2.5 (9)   | C15—C16—C17—C18 | 1.9 (10)   |
| C2—C3—C8—C7  | 176.8 (6)  | Br1—C16—C17—C18 | -176.2 (5) |
| C4—C3—C8—C9  | 178.3 (6)  | C16—C17—C18—C19 | -2.2 (10)  |
| C2—C3—C8—C9  | -2.4 (9)   | C17—C18—C19—C14 | 2.0 (10)   |
| C7—C8—C9—C10 | -178.3 (6) | C15—C14—C19—C18 | -1.5 (10)  |
| C3—C8—C9—C10 | 0.9 (10)   | C13—C14—C19—C18 | 174.6 (6)  |
| C8—C9—C10—C1 | 0.4 (10)   | C2—C1—O1—C20    | 1.2 (9)    |
| C2—C1—C10—C9 | -0.2 (10)  | C10—C1—O1—C20   | -177.5 (6) |
| O1—C1—C10—C9 | 178.6 (6)  |                 |            |

*Hydrogen-bond geometry* ( $\text{\AA}$ ,  $^\circ$ )

Cg2 is the centroid of the C3—C8 ring.

| $D-H\cdots A$                    | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|----------------------------------|-------|-------------|-------------|---------------|
| C4—H4 $\cdots$ Cg2 <sup>i</sup>  | 0.95  | 2.70        | 3.432 (6)   | 134           |
| C7—H7 $\cdots$ Cg2 <sup>ii</sup> | 0.95  | 2.80        | 3.520 (6)   | 134           |

Symmetry codes: (i)  $-x+1/2, y-3/2, z$ ; (ii)  $-x+1, y+1/2, -z+1/2$ .